organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Silvia Stumpf, Matthias Wagner and Michael Bolte*

Institut für Anorganische Chemie, J. W. Goethe-Universität Frankfurt, Marie-Curie-Str. 11, 60439 Frankfurt/Main, Germany

Correspondence e-mail: bolte@chemie.uni-frankfurt.de

Key indicators

Single-crystal X-ray study T = 173 K Mean σ (C–C) = 0.007 Å Disorder in main residue R factor = 0.076 wR factor = 0.210 Data-to-parameter ratio = 17.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

4-Hydroxymethyl-3,5-dimethyl-1*H*-pyrazole chloroform solvate

In the title compound, $C_6H_{10}N_2O \cdot CHCl_3$, the geometric parameters are in the usual ranges. The structure is stabilized by $N-H \cdot \cdot \cdot O$, $O-H \cdot \cdot \cdot N$ and $C-H \cdot \cdot \cdot O$ hydrogen bonds.

Received 7 September 2004 Accepted 8 September 2004 Online 18 September 2004

Comment

A perspective view of the title compound, (I), is shown in Fig. 1. The structure is composed of discrete 3,5-dimethyl-4-hydroxymethylpyrazole and chloroform solvent molecules. Bond lengths and angles can be regarded as normal (Cambridge Crystallographic Database, *CONQUEST* Version 1.6 plus three updates, *MOGUL* Version 1.0; Allen, 2002). The pyrazole heterocycle is essentially planar. The hydroxyl group is rotated by -78.1 (6)° out of this plane. The structure is stabilized by N-H···O, O-H···N and C-H···O hydrogen bonds (Fig. 2 and Table 2).



Experimental

3,5-Dimethyl-4-hydroxymethylpyrazole was synthesized in the following way (Stumpf, 1999): 4-bromo-3,5-dimethylpyrazole was treated with five equivalents of *tert*-butyllithium in THF. Four equivalents of DMF were added and the resulting 3,5-dimethyl-4-formylpyrazole reduced with sodium borohydride in ethanol to the corresponding alcohol.

Crystal data	
$\begin{aligned} &C_{6}H_{10}N_{2}O\cdot CHCl_{3} \\ &M_{r} = 245.53 \\ &Monoclinic, \ P2_{1}/c \\ &a = 12.423 \ (2) \ \text{\AA} \\ &b = 7.843 \ (1) \ \text{\AA} \\ &c = 12.871 \ (2) \ \text{\AA} \\ &\beta = 109.44 \ (2)^{\circ} \\ &V = 1182.6 \ (3) \ \text{\AA}^{3} \\ &Z = 4 \end{aligned}$	$D_x = 1.379 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 1700 reflections $\theta = 1-25^{\circ}$ $\mu = 0.74 \text{ mm}^{-1}$ T = 173 (2) K Plate, colourless $0.40 \times 0.30 \times 0.05 \text{ mm}$
Data collection	
Siemens SMART CCD three-circle diffractometer ω scans Absorption correction: multi-scan (<i>SADABS</i> ; Sheldrick, 1996) $T_{\min} = 0.756, T_{\max} = 0.964$ 8134 measured reflections	2084 independent reflections 1067 reflections with $I > 2\sigma(I)$ $R_{int} = 0.074$ $\theta_{max} = 25.0^{\circ}$ $h = -14 \rightarrow 14$ $k = -9 \rightarrow 9$ $l = -15 \rightarrow 15$

© 2004 International Union of Crystallography Printed in Great Britain – all rights reserved



Figure 1

Perspective view of the title compound, with the atom numbering. Displacement ellipsoids are drawn at the 50% probability level.

Refinement

$w = 1/[\sigma^2(F_o^2) + (0.0872P)^2]$
+ 1.7867P]
where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} < 0.001$
$\Delta \rho_{\rm max} = 0.43 \ {\rm e} \ {\rm \AA}^{-3}$
$\Delta \rho_{\rm min} = -0.59 \text{ e } \text{\AA}^{-3}$

Table 1

Selected geometric parameters (Å, °).

N1-C5	1.335 (6)	C3-C4	1.390 (7)
N1-N2	1.344 (6)	C4-C5	1.405 (7)
N2-C3	1.349 (6)		
C5-N1-N2	105.3 (4)	C3-C4-C5	105.5 (4)
N1-N2-C3	113.0 (4)	N1-C5-C4	110.6 (4)
N2-C3-C4	105.7 (4)		

Table 2

Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdots A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$N2-H2\cdots O41^{i}$	0.88	1.97	2.830 (5)	166
$C1 - H1 \cdots O41$	1.00	2.14	3.094 (7)	159
$O41 - H41 \cdots N1^{ii}$	0.84	1.88	2.707 (5)	167
	1 . 1	(**) 3	1	

Symmetry codes: (i) $1 - x, \frac{1}{2} + y, \frac{1}{2} - z$; (ii) $x, \frac{3}{2} - y, z - \frac{1}{2}$.



Figure 2

Packing diagram of the title compound, viewed on the *bc* plane. Colour codes: C: shaded black circles; H: small white circles; Cl: cross-hatched green circles; N: dotted blue circles; O: shaded red circles.

H atoms were refined with fixed individual displacement parameters [U(H) = 1.2 U_{eq} (C, N, O) or U(H) = 1.5 U_{eq} (C_{methyl})] using a riding model, with O–H = 0.84 Å, N–H = 0.88 Å and C–H ranging from 0.98 to 1.00 Å. One of the methyl groups was found to be disordered over two positions. The site occupation factor for the two different orientations refined to 0.45 (6)/0.55 (6). The hydroxyl group and the non-disordered methyl group were allowed to rotate but not to tip.

Data collection: *SMART* (Siemens, 1995); cell refinement: *SAINT* (Siemens, 1995); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1990); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *XP* in *SHELXTL* (Sheldrick, 1991); software used to prepare material for publication: *SHELXL*97 and *PLATON* (Spek, 2003).

References

- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1991). *SHELXTL-Plus*. Release 4.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Siemens (1995). *SMART* and *SAINT*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
- Stumpf, S. (1999). Diploma thesis, Universität Frankfurt, Germany.